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Design, Synthesis, and Structure—Activity Relationship of a Novel Series of 2-Aryl 5-(4-Oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans as HIV-1 Entry Inhibitors¹

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We previously identified two small molecules targeting the HIV-1 gp41, N-(4-carboxy-3-hydroxy)phenyl-2,5-dimethylpyrrole 12 (NB-2) and N-(3-carboxy-4-chloro)phenylpyrrole 13 (NB-64), that inhibit HIV-1 infection at low micromolar levels. On the basis of molecular docking analysis, we designed a series of 2-aryl 5-(4-oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans. Compared with 12 and 13, these compounds have bigger molecular size (437–515 Da) and could occupy more space in the deep hydrophobic pocket on the gp41 NHR trimer. Fifteen 2-aryl 5-(4-oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans (11a–0) were synthesized by Suzuki–Miyaura cross-coupling followed by a Knoevenagel condensation and tested for their anti-HIV-1 activity and cytotoxicity on MT-2 cells. We found that all 15 compounds had improved anti-HIV-1 activity and 3 of them (11a, 11b, and 11d) exhibited inhibitory activity against replication of HIV-1_{IIIB} and 94UG103 at < 100 nM range, more than 20-fold more potent than 12 and 13, suggesting that these compounds can serve as leads for development of novel small molecule HIV fusion inhibitors.

Introduction

So far, over 60 million people have been infected with HIV and of these more than 25 million have died of AIDS. ^{1,2} As of March of 2009, the U.S. Food and Drug Administration (FDA) has licensed 28 anti-HIV drugs, including 15 reverse transcriptase inhibitors (RTI^a) and 10 protease inhibitors (PI) (http://www.hivandhepatitis.com/hiv_and_aids/hiv_treat.html). Application of these antiretroviral drugs in combination, designated as highly active antiretroviral therapy (HAART), has led to significant reduction of mobility and mortality of HIV/AIDS. However, an increasing number of HIV/AIDS patients fail to respond to current antiretroviral therapeutics because of the emergence of multidrug resistance to RTIs and PIs and serious adverse side effects. ⁴⁻⁶ Therefore, there is an urgent need to develop novel anti-HIV therapeutics targeting different steps of HIV replication cycle, particularly the HIV fusion and entry events.

HIV type 1 (HIV-1) entry into the host cell is initiated by binding of its envelope glycoprotein (Env) surface subunit

gp120 to the CD4 molecule and a coreceptor, CCR5 or CXCR4, causing a series of conformational changes of the Env transmembrane subunit gp41.⁷ The fusion peptide located at the N-terminus of gp41 inserts into the target cell membrane, followed by association between the N- and C-terminal heptad repeats (NHR and CHR, respectively) to form a six-helix bundle core structure, bringing the viral envelope and target cell membrane into proximity. 8,9 Peptides derived from the gp41 CHR region can interact with the viral gp41 NHR region and block fusogenic core formation, resulting in potent inhibition of HIV-1 fusion with the target cells. ^{10–14} One of the CHR-peptides, enfuvirtide (T-20), was licensed by the U.S. FDA as the first member of a new class of anti-HIV drugs, HIV fusion inhibitors for treating HIV/AIDS patients who have failed to respond to RTI and PI. 15,16 However, clinical application of the enfuvirtide is limited because of its lack of oral availability, induction of drug resistance, and high cost of production. Therefore, it is critical to develop orally available non-peptide small-molecule fusion inhibitors.

Using a fluorescence-linked immunoabsorbance assay with a conformation-specific monoclonal antibody (mAb) NC-1, ¹⁷ we previously identified two small molecules, *N*-(4-carboxy-3-hydroxy)phenyl-2,5-dimethylpyrrole **12** (NB-2) and *N*-(3-carboxy-4-chloro)phenylpyrrole **13** (NB-64), which inhibit HIV-1 fusion and gp41 six-helix bundle formation at low micromolar levels (Figure 1). ¹⁸ Molecular docking analysis of these two compounds indicates that **12** and **13** each occupy only a part of the deep hydrophobic pocket on the gp41 NHR trimer. ¹⁸ We therefore reasoned that new lead compounds with improved anti-HIV-1 potency could be designed by increasing the molecular size so that they would occupy more space in the hydrophobic pocket. We now report the design and synthesis of 15 derivatives of 2-aryl

¹ Dedicated to Professor Corwin Hansch on his 91st anniversary in appreciation of his pioneering achievements.

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[&]quot;Abbreviations: CHR, C-terminal heptad repeat; Env, envelope glycoprotein; NHR, N-terminal heptad repeat; XTT, sodium 3'-[1-(phenylamino)-carbonyl]-3,4-tetrazolium-bis(4-methoxy-6-nitro)bezenesulfonic acid hydrate; RTI, reverse transcriptase inhibitors; PI, protease inhibitors; HAART, highly active antiretroviral therapy; mAb, monoclonal antibody; ELISA, enzymelinked immunosorbent assay; THF, tetrahydrofuran; LDA, lithium diisopropylamide; i-PrMgCl, isopropylmagnesium chloride; EtOH–DME, ethyl alcohol—ethylene glycol dimethyl ether; RCSB, Research Collaboratory for Structural Bioinformatics; MOI, multiplicity of infection; PMS, phenazine methosulphate.

5-(4-oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furan (11a-o) with molecular weight ranging from 437 to 515 Da. We evaluated their antiviral activity against HIV-1_{IIIB} and their cytotoxicity on MT-2 cells that were used in the viral infection assay. Strikingly, all 15 compounds exhibited improved anti-HIV-1 activity and 3 of them (11a, 11b, and 11d) inhibited replication by laboratory-adapted and primary HIV-1 strains with EC₅₀ (effective concentration for 50% inhibition) ranging from 44 to 99 nM and SI (selectivity index) in the range of 330–440. These results suggest that 2-aryl 5-(4-oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans show great potential for further development as novel small molecule anti-HIV therapeutics targeting gp41.

Results and Discussion

A total of 15 2-aryl 5-(4-oxo-3-phenethyl-2-thioxothiazoli-dinylidenemethyl)furans (11a-o) were synthesized by Suzuki-Miyaura cross-coupling, followed by a Knoevenagel condensation. Their molecular weights ranged from 436 to 515 Da.

We tested the inhibitory activity of these compounds on the replication of the laboratory-adapted HIV-1 strain IIIB, a prototype of the X4 strain, in MT-2 cells that express CD4 and the coreceptor CXCR4 by measuring p24 production in an enzyme-linked immunosorbent assay (ELISA) as previously described. ¹⁸ All 15 compounds significantly inhibited HIV-1 replication in a dose-dependent manner with the EC₅₀ (effective concentration for 50% inhibition) ranging from 0.042 to 1.3 μ M. We also tested the cytotoxicity of these compounds on MT-2 cells using an XTT {sodium 3'-[1-(phenylamino)-carbonyl]-3,4-tetrazolium-bis(4-methoxy-6-nitro)bezenesulfonic acid hydrate} assay. ¹⁹ The CC₅₀ (concentration causing 50% cytotoxicity) values ranged from 3.29 to 78.29 μ M, and their selectivity indexes (SI) ranged from 10 to 915 (Table 3 and

HO COOH HOOC
$$H_3$$
C H_3 C H

Figure 1. Chemical structures of 12 and 13.

and the worst compound is 11f, with an EC₅₀ of 1.28 μ M and a SI of 12. Although compound 11i had an appreciable potency (EC₅₀ = 0.32 μ M), it had high cytotoxicity (CC₅₀ = 3.29 μ M), resulting in a SI of 10.

Four possible isomers need to be considered for compounds 11a-o: A, B, C, and D (Figure 2). The interconvertions A \leftrightarrow B and C \leftrightarrow D should be easy as they proceed by rotation about single bonds. The interconversions of A \leftrightarrow C and B \leftrightarrow D will be

Figure S1 of Supporting Information). The best compound is

111, which has the lowest EC₅₀ (42 nM) and highest SI (915),

Tour possible isomers need to be considered for compounds 11a-o: A, B, C, and D (Figure 2). The interconvertions A↔B and C↔D should be easy as they proceed by rotation about single bonds. The interconversions of A↔C and B↔D will be more difficult because they need rotation about double bonds. We also expect that isomers A and B will be more stable than C and D because of the cross space interactions. To examine the possible intervention of the unstable product isomers C and D, we tested the anti-HIV-1_{IIIB} activity of these compounds under exclusion of light. As shown in Table 3, there was no substantial difference in antiviral activity of the compounds tested under dark and light conditions. This excludes the possibility that light-activation to give isomers C and D is the origin for their bioactivity.

We also tested the inhibitory activity of these compounds on the replication of a representative primary HIV-1 isolate 94UG103 (R5X4, clade A) in PBMCs. All 15 compounds effectively inhibited HIV-1 94UG103 replication similarly in a dose-dependent manner with the EC₅₀ ranging from 0.07 to 4 μ M (Table 3 and Figure S1). The best compounds are 11a, 11b, and 11d, which have the lowest EC₅₀ (70–74 nM). This result suggests that these compounds are much more potent than 12 and 13 in inhibiting infection by primary HIV-1. Molecular docking analysis revealed that the phenethyl group of compound 11d filled the space in the deep hydrophobic pocket of gp41 formed by the NHR trimer (Figure 3), previously observed to be unoccupied by 13.

This may explain why 11d has \sim 25-fold greater potency than 13. In the current series we kept the phenethyl group fixed and substituted the phenyl ring attached to the furan ring to derive a comprehensive structure—activity relationship (SAR). We showed previously that the carboxylic acid group in this phenyl ring is essential for anti-HIV-1 activity, and hence, it was retained throughout. It appears that the compound 11a with no substituent at the 2 and 6 positions of the phenyl ring showed highly potent activity with reasonably good selectivity index. Addition of hydrophobic substituent at R (11b-d) had moderate effects and reduced the anti-HIV-1 activity by about 2- to 4-fold. However, bromine substitution

Figure 2. Possible E-Z isomers of compounds 11a-o.

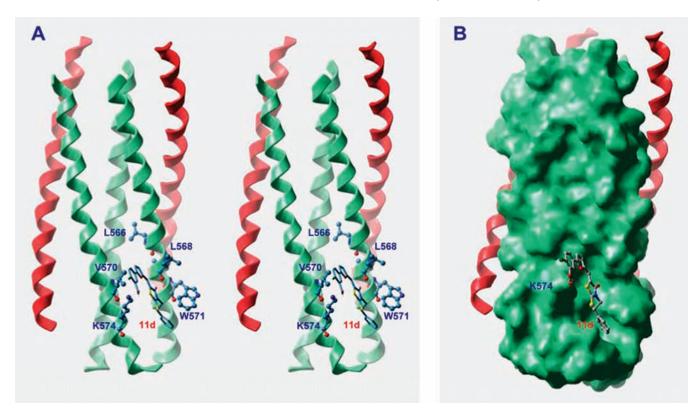


Figure 3. Docking of 11d in the gp41 hydrophobic cavity. (A) The stereoview of 11d docked in the hydrophobic cavity showing possible interactions with the neighboring hydrophobic and charged residue K574. (B) Surface representation of the gp41 core with bound ligand 11d. The compound docked inside the cavity. The negatively charged COOH group is pointing toward the positively charged area contributed by K574.

Scheme 1. Synthesis of 2-Ethyl-5-bromobenzoic Acid $2c^a$

^a Reagents and conditions: (a) Br₂/iron (powder), room temperature to 50 °C, 14%.

(11i) had a more severe effect probably due to steric effects and the activity dropped by \sim 7-fold. The most marked adverse effect on activity was demonstrated by electron donating substituents OH and OCH₃, where the anti-HIV-1 activity dropped by \sim 30-fold. Introduction of a nitrogen atom in the ring in the unsubstituted 11a yielded the most active inhibitor (111 EC₅₀ = 42 nm, but when chlorine or bromine was substituted at the R position (11m and 11n) or a methyl group was introduced at the R' position (110), a substantial decrease in activity was observed. Interestingly moving the methyl group of 11b to the R' position yielded 11h, which was more active than 11b. However, similar moves for chlorine and fluorine had marginal effects. This SAR analysis will help us in further modifying the most active analogues obtained from the current study.

In the next stage, we will further evaluate the inhibitory activities of these 15 compounds on the HIV-1-mediated cell-cell fusion and cytopathic effect as well as the gp41 sixhelix bundle formation. These data in combination with anti-HIV-1 activity and cytotoxicity well refine the structure—activity relationship, which will be used to design new, small molecule

Scheme 2. Synthesis of 3-Bromo-2-chlorobenzoic Acid 2j^a

^a Reagents and conditions: (a) KMnO₄, H₂O, reflux, 24 h, 49%.

HIV-1 fusion inhibitors with improved anti-HIV-1 potency and reduced cytotoxicity for further preclinical evaluation.

Chemistry

Synthesis of Aryl(heteryl) Bromide/Iodides 2c, 2j, 2n, 2o. 2-Ethyl-5-bromobenzoic acid **2c** was isolated in 14% yield by treatment of 2-ethylbenzoic acid 1c with bromine in the presence of iron powder at room temperature for 48 h and at 55 °C for 2 h (Scheme 1).

3-Bromo-2-chlorobenzoic acid 2j³³ was synthesized in 49% yield by the oxidation of 3-bromo-2-chlorotoluene with KMnO₄ in water for 24 h at reflux temperature (Scheme 2).

2-Bromo-5-iodonicotinic acid **2n** was synthesized by treatment of 2,5-dibromo-3-picoline **1n** with isopropylmagnesium chloride in the presence of I₂ in tetrahydrofuran (THF) at room temperature for 24 h, followed by oxidation³⁴ using KMnO₄ in water for 48 h at reflux temperature (Scheme 3).

4-Methyl-5-iodonicotinic acid 20 was synthesized from 3bromo-5-(4,4-dimethyl-1,3-oxazolin-2-yl)pyridine 10. First, 3-bromo-5-(4,4-dimethyl-1,3-oxazolin-2-yl)pyridine 10²⁰ was treated with MeI in the presence of lithium diisopropylamide (LDA) at -78 °C for 1 h and for 1 h at room temperature to afford 3-bromo-4-methyl-5-(4,4-dimethyl-1,3-oxazolin-2-yl)-pyridine **30** in 76% yield. Reaction of 3-bromo-4-methyl-5-(4,4-dimethyl-1,3-oxazolin-2-yl)pyridine **30** with isopropyl-magnesium chloride in the presence of I_2 in THF at room temperature for 24 h followed by hydrolysis gave 4-methyl-5-iodonicotinic acid **20** (Scheme 4).

Synthesis of 3-Phenethyl-2-thioxothiazolidin-4-one 6. 3-Phenethyl-2-thioxothiazolidin-4-one **6** was synthesized by modification of a literature procedure. ²¹ Phenylethylamine **5**

Scheme 3. Synthesis of 2-Bromo-5-iodonicotinic Acid 2n^a

^a Reagents and conditions: (a) *i*-PrMgCl, I₂, THF, room temp, 24 h, 55%; (b) KMnO₄, H₂O, reflux, 48 h, 14%.

Scheme 4. Synthesis of 4-Methyl-5-iodonicotinic Acid 20^a

 a Reagents and conditions: (a) LDA, MeI, -78 °C, 1 h, 76%; (b) *i*-Pr-MgCl, I₂, THF, room temp, 24 h, 45%; (c) 5 N HCl, reflux, 24 h, 84%.

Scheme 5. Synthesis of 3-Phenethyl-2-thioxothiazolidin-4-one 6^a

 a Reagents and conditions: (a) (i) CS₂, Et₂O, 0–5 °C, 0.5 h; (ii) ClCH₂-COOH, EtOH, 1 h, reflux, 56%.

was treated with CS_2 in diethyl ether at 0-5 °C followed by treatment with chloroacetic acid under reflux for 1 h to give 3-phenethyl-2-thioxothiazolidin-4-one **6** in 56% yield (Scheme 5).

Synthesis of 2-Aryl 5-Formylfurans 8. 2-Aryl 5-formylfurans 8 were synthesized by palladium-catalyzed Suzuki—Miyaura cross-coupling 22,23 of aryl(heteryl) bromide/iodide with aryl(heteryl) boronic acids. 2-Aryl 5-formylfurans 8a-o were obtained (i) by coupling of 5-formyl-2-furylboronic acid 7 with corresponding aryl(heteryl) bromides/iodides 2b-d,f-o to give 8b-d,f-o (28-80% yield) by a modified literature method of (ii) by reaction of 5-bromofuran-2-carboxaldehyde 9 with 3-carboxyphenylboronic acid 10a,e by a slight modifications to Bumagin's method of 5-bromofuran coupling did not occur when the bromo analogues of 2i, 2n, 2o were used instead of aryl(heteryl) iodides 2i, 2n, 2o. All new compounds were characterized by 1 H and 13C NMR spectroscopy and elemental analysis.

Synthesis of 2-Aryl 5-(4-Oxo-3-phenethyl-2-thioxothiazolidiny-lidenemethyl)furans 11. Knoevenagel condensation $^{26-28}$ of N-substituted rhodanine 6 with aldehydes $8\mathbf{a}-\mathbf{o}$ in ethanol in the presence of a catalytic amount of 2,2,6,6-tetramethylpiperidine at 78 °C for 1.5–12.5 h gave $11\mathbf{a}-\mathbf{o}$ (38–76%) (Scheme 7, Table 2). Novel 2-aryl 5-(4-oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans $11\mathbf{a}-\mathbf{o}$ were characterized by 1 H and 13 C NMR spectroscopy and elemental analysis.

Conclusion. In conclusion, we have designed and synthesized a series of novel 2-aryl 5-(4-oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans **11a**–**0** by Suzuki–Miyaura cross-coupling, followed by Knoevenagel condensation. These molecules

Table 1. Preparation of 2-Aryl 5-Formylfurans 8

product 8	X of 2, 8, and 10	R of 2, 8, and 10	R ¹ of 2 , 8 , and 10	reaction time, h	yield (%) of 8 ^a
8a	СН	Н	Н	1.5	89
8b	CH	CH_3	H	8.0	60
8c	CH	C_2H_5	H	7.0	68
8d	CH	Cl	H	15	51 ^b
8e	CH	F	H	1.5	59
8f	CH	OH	H	7.0	67
8g	CH	OCH_3	H	3.5	52
8h	CH	Н	CH_3	1.0	80
8i	CH	Br	H	16.0	47
8j	CH	H	Cl	16.0	49
8k	CH	H	F	7.5	74
81	N	Н	H	28.0	40
8m	N	Cl	H	28.0	28
8n	N	Br	H	16.0	31
80	N	Н	CH ₃	16.0	56

^a Isolated yields. ^b Crude yield and used without purification.

Scheme 6. Synthesis of 2-Aryl 5-Formylfurans $8^{a,b}$

OOH

$$R + R^1$$
 $R + R^1$
 $R +$

^b For designation of X, R and R¹ in 2, 8 and 10 see Table 1.

^a Reagents and conditions: (a) (PPh₃)₂PdCl₂, Na₂CO₃ (aq), EtOH–DME (1:1), 50 °C, 28–80%; (b) Pd(OAc)₂, Na₂CO₃, H₂O, room temp, 59–89%.

Scheme 7. Synthesis of 2-Aryl 5-(4-Oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans 11^a

^a Reagents and conditions: (a) ethanol, cat. 2,2,6,6-tetramethylpiperidine, reflux, 38–76%. For designation of X, R, and R¹ in 8 and 11, see Tables 1 and 2.

Table 2. Preparation of 2-Aryl 5-(4-Oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans 11a-o

product 11	X of 11	R of 11	R ¹ of 11	reaction time, h	yield (%) of 11 ^a
11a	СН	Н	Н	1.5	52
11b	CH	CH_3	H	1.5	57
11c	CH	C_2H_5	H	3.0	48
11d	CH	Cl	Н	3.0	60
11e	CH	F	H	1.5	71
11f	CH	OH	H	12.5	50
11g	CH	OCH_3	Н	2.0	56
11h	CH	Н	CH_3	1.0	38
11i	CH	Br	Н	6.0	54
11j	CH	Н	Cl	9.0	58
11k	CH	Н	F	5.0	65
111	N	Н	Н	2.0	48
11m	N	C1	Н	3.0	49
11n	N	Br	Н	10.0	74^{b}
11o	N	Н	CH_3	10.0	76

^a Isolated yield. ^b One equivalent of 2,2,6,6-tetramethylpiperidine was used.

exhibited highly potent anti-HIV-1 activity. The most promising compounds are 11a, 11b, and 11d, which have the lowest EC₅₀ (70–74 nM) and which inhibited replication by laboratoryadapted and primary HIV-1 strains at < 100 nM with selectivity indexes ranging from 330 to 440, suggesting that these compounds can be used as a lead for developing novel small molecule HIV fusion/entry inhibitors for treatment of HIV/AIDS patients who have failed to respond to current antiretroviral therapeutics.

Experimental Section

Chemistry. Arvl(heteryl) bromides/iodides 2b, 2d-i, 2k-m, 9 and arylboronic acids 10a, 10e were obtained from commercial suppliers and used without further purification. Aryl(heteryl) bromides/iodides 2c, 2j, 2n, 2o and 3-phenethyl-2-thioxothiazolidin-4-one 6 were synthesized, and complete details of the synthesis and spectral data are given in the Supporting Information. 5-Formyl-2-furanboronic acid 7 was synthesized according to a literature method.²⁹ Melting points were determined using a capillary melting point apparatus equipped with a digital thermometer and are uncorrected. ¹H (300 MHz) and ¹³C (75 MHz) NMR spectra were recorded in DMSO-d₆ (with tetramethylsilanes as the internal standard), unless otherwise stated. Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, br s = broad, m = multiplet), coupling constants (J values) in Hz. Elemental analyses were performed on a Carlo Erba EA-1108 instrument. All the reactions were performed in flame-dried glassware. The solvents (ethylene glycol dimethyl ether (DME) and THF) were dried by the usual methods and distilled before use. Purity of compounds was determined by elemental analyses and/or HPLC; purity of target compounds was ≥95% unless otherwise noted. Analytical HPLC analyses were performed on Shimadzu SPD-20-A

using a Whelk-O1 column with detection at 254 nm, a flow rate of 1.0 mL/min, and methanol as the eluting solvent. Column chromatography was performed on silica gel 200-425 mesh.

General Procedure for Preparation of 3-(5-Formylfuran-2yl)benzoic Acids (8a and 8e). 5-Bromofuran-2-carbaldehyde 9 (0.44 g, 2.5 mmol), sodium carbonate (0.79 g, 7.5 mmol), and 3-carboxyphenylboronic acid 10 (2.75 mmol) were suspended in degassed water (12.5 mL) under nitrogen. Palladium acetate (5.6 mg, 0.025 mmol) was added to this mixture, which was stirred at room temperature for 1.5 h. The slurry was dissolved in water (100 mL) and filtered through Celite. The filtrate was acidified with HCl and the precipitate was filtered and washed with water to give the corresponding 5-(5-formylfuran-2yl)benzoic acids 8a and 8e. Analytical samples were obtained by recrystallization from alcohol.

3-(5-Formylfuran-2-yl)benzoic Acid (8a). Gray microcrystals, mp 264-266 °C (2-propanol) (lit. 30 mp 266-267 °C); ¹H NMR (DMSO- d_6) δ 9.63 (s, 1H), 8.37 (br s, 1H), 8.12 (d, J = 8.0 Hz, 1H), 7.99 (d, J = 7.9 Hz, 1H), 7.68–7.62 (m, 2H), 7.41 (d, J =3.7 Hz, 1H); 13 C NMR (DMSO- d_6) δ 178.3, 166.9, 157.2, 152.0, 131.9, 130.3, 129.8, 129.3, 129.1, 125.4, 109.7, 108.3. Anal. Calcd for C₁₂H₈O₄: C, 66.67; H, 3.73. Found: C, 66.40; H, 3.65.

2-Fluoro-5-(5-formylfuran-2-yl)benzoic Acid (8e). White microcrystals, mp 231–233 °C (ethanol); ¹H NMR (DMSO- d_6) δ 9.64 (s, 1H), 8.31 (dd, J = 6.9, 2.4 Hz, 1H), 8.17 - 8.12 (m, 1H),7.68 (d, J = 3.7 Hz, 1H), 7.50 (dd, J = 10.5, 8.8 Hz, 1H), 7.40 (d, J = 10.5, 8.8 Hz, 1H), 7 $J = 3.7 \text{ Hz}, 1\text{H}; ^{13}\text{C NMR (DMSO-}d_6) \delta 178.1, 164.5 (d, <math>J =$ 3.4 Hz, 1C), 161.5 (d, J = 261.1 Hz, 1C), 156.3, 151.9, 131.2, (d, J = 9.7 Hz, 1C, 128.2, 125.3 (d, J = 4.0 Hz, 1C), 120.3 (d, J = 4.0 Hz, 1C)11.5 Hz, 1C), 118.3 (d, J = 23.5 Hz, 1C), 109.4. Anal. Calcd for C₁₂H₇FO₄: C, 61.54; H, 3.01. Found: C, 61.35; H, 3.24.

General Procedure for the Preparation of 3-(5-Formylfuran-2yl)benzoic and -nicotinic Acids (8b-d,f-o). A mixture of aryl-(heteryl) bromides/iodides 2b-d,f-o (2.3 mmol), 5-formyl-2-furanboronic acid 7 (0.42 g, 3.0 mmol), and (Ph₃P)₂PdCl₂ (84.2 mg, 0.12 mmol) in DME (7.0 mL), ethanol (7.0 mL), and 2 M aqueous Na₂CO₃ (6.9 mL, 13.8 mmol of Na₂CO₃) was flushed with nitrogen for 5 min and heated at 50 °C for 3.5-28 h (reaction time is given in Table 1) under nitrogen atmosphere. The solvents were removed under reduced pressure, the residue was dissolved in water (20 mL), the mixture obtained was filtered through Celite, and the filtrate was neutralized with 2 N hydrochloric acid. The solids were filtered, washed with water, dried, and recrystallized from ethanol to give 3-(5-formylfuran-2-yl)benzoic and -nicotinic acids 8b-d,f-o.

5-(5-Formylfuran-2-yl)-2-methylbenzoic acid (8b). Brownish microcrystal, mp 236–238 °C (ethanol); ¹H NMR (DMSO-d₆) δ 9.63 (s, 1H), 8.28 (d, J = 1.8 Hz, 1H), 7.95 (dd, J = 7.8, 1.8 Hz, 1H), 7.67 (d, J = 3.7 Hz, 1H), 7.46 (d, J = 8.1 Hz, 1H), 7.35 (d, $J = 3.7 \text{ Hz}, 1\text{H}), 2.58 \text{ (s, 3H)}; ^{13}\text{C NMR (DMSO-}d_6) \delta 177.9,$ 168.1, 157.3, 151.7, 140.7, 132.6, 131.4, 128.0, 126.5, 126.3, 125.4, 108.9, 21.2. Anal. Calcd for C₁₃H₁₀O₄: C, 67.82; H, 4.38. Found: C, 67.60; H, 4.47.

5-(5-Formylfuran-2-yl)-2-ethylbenzoic Acid (8c). Brown prisms, mp 161–163 °C; ¹H NMR (DMSO- d_6) δ 9.63 (s, 1H),

915

165

40

202

 $EC_{50} (\mu M)$ MW HIV-1_{IIIB}^b HIV-1_{IIIB} 94UG103^d SI^e product $CC_{50}(\mu M)$ 435.5 0.044 ± 0.005 0.098 ± 0.033 0.074 ± 0.015 19.35 ± 0.75 440 11a 11b 449.5 0.099 ± 0.010 0.031 ± 0.016 0.070 ± 0.012 42.14 ± 1.61 426 463.6 0.081 ± 0.011 0.030 ± 0.032 0.177 ± 0.061 17.39 ± 0.90 215 11c 0.051 ± 0.013 0.017 ± 0.003 0.073 ± 0.063 16.82 ± 1.03 330 11d 470.0 380 11e 453.5 0.071 ± 0.010 0.291 ± 0.241 0.246 ± 0.037 27.01 ± 1.48 11f 451.5 1.280 ± 0.240 0.261 ± 0.098 2.660 ± 0.610 14.78 ± 1.99 12 465.6 1.300 ± 0.151 0.266 ± 0.033 2.229 ± 0.176 28.72 ± 1.70 22 11g 11h 436.5 0.043 ± 0.008 0.092 ± 0.004 0.648 ± 0.108 26.84 ± 3.73 624 0.165 ± 0.019 3.29 ± 1.05 10 11i 514.4 0.317 ± 0.071 0.158 ± 0.046 11j 470.0 0.064 ± 0.005 0.078 ± 0.017 0.033 ± 0.010 6.73 ± 0.56 105 11k 453.5 0.094 ± 0.002 0.059 ± 0.014 0.801 ± 0.138 78.29 ± 0.57 835

Table 3. Anti-HIV-1 Activity and Selectivity Indexes of the 2-Aryl 5-(4-Oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl) furans $11a - o^a$

 a Each compound was tested in triplicate; the data are presented as the mean \pm SD. b The experiment was performed without exclusion of light. c The experiment was performed with exclusion of light. d 94UG103 is a primary HIV-1 isolate (X4R5, clade A). c SI was calculated on the basis of the CC₅₀ for MT-2 cells and EC₅₀ for inhibiting infection of HIV-1_{IIIB}.

 0.071 ± 0.004

 1.010 ± 0.310

 0.227 ± 0.080

 0.538 ± 0.045

8.23 (d, J = 1.9 Hz, 1H), 7.97 (dd, J = 8.0, 1.9 Hz, 1H), 7.67 (d, J = 3.7 Hz, 1H), 7.48 (d, J = 8.1 Hz, 1H), 7.35 (d, J = 3.7 Hz, 1H), 2.97 (q, J = 7.4 Hz, 2H), 1.19 (t, J = 7.4 Hz, 3H); ¹³C NMR (DMSO- d_6) δ 177.9, 168.3, 157.4, 151.7, 146.4, 131.4, 131.2, 128.0, 126.4, 126.3, 125.5, 109.0, 26.7, 15.8. Anal. Calcd for $C_{14}H_{12}O_4$: C, 68.85; H, 4.95. Found: C, 68.51; H, 5.04.

 0.042 ± 0.019

 0.232 ± 0.020

 0.268 ± 0.055

 0.390 ± 0.070

449.6

471.0

515.4

450.5

111

11m

11n

11₀

5-(5-Formylfuran-2-yl)-2-hydroxybenzoic Acid (8f). Brown microcrystals, mp 245–247 °C; ¹H NMR (DMSO- d_6) δ 9.59 (s, 1H), 8.24 (d, J = 2.2 Hz, 1H), 8.03 (dd, J = 8.8, 2.3 Hz, 1H), 7.65 (d, J = 3.7 Hz, 1H), 7.23 (d, J = 3.8 Hz, 1H), 7.11 (d, J = 8.8 Hz, 1H); ¹³C NMR (DMSO- d_6) δ 177.5, 171.3, 162. 0, 157.6, 151.4, 132.3, 126.7, 125.8, 120.1, 118.3, 113.8, 107.7. Anal. Calcd for $C_{12}H_8O_5$: C, 62.08; H, 3.47. Found: C, 61.67; H, 3.49.

5-(5-Formylfuran-2-yl)-2-methoxybenzoic Acid (8g). Yellowish microcrystals, mp 220–222 °C; ¹H NMR (DMSO- d_6) δ 9.59 (s, 1H), 8.12 (d, J = 2.3 Hz, 1H), 8.02 (dd, J = 8.8, 2.3 Hz, 1H), 7.65 (d, J = 3.9 Hz, 1H), 7.30–7.26 (m, 2H), 3.90 (s, 3H); ¹³C NMR (DMSO- d_6) δ 177.6, 166.8, 159.0, 157.6, 151.4, 129.7, 127.3, 125.8, 122.1, 120.8, 113.3, 107.9, 56.1. Anal. Calcd for $C_{13}H_{10}O_5$: C, 63.42; H, 4.09. Found: C, 63.22; H, 4.04.

3-(5-Formylfuran-2-yl)-2-methylbenzoic Acid (**8h**). Yellow microcrystals, mp 140–142 °C; ¹H NMR (DMSO- d_6) δ 9.71 (s, 1H), 8.01 (d, J = 7.8 Hz, 1H), 7.89 (d, J = 7.8 Hz, 1H), 7.42–7.36 (m, 2H), 6.75 (d, J = 3.6 Hz, 1H), 2.74 (s, 3H); ¹³C NMR (DMSO- d_6) δ 177.1, 169.9, 158.3, 151.5, 136.7, 133.1, 131.4, 130.9, 130.1, 125.3, 112.0, 18.3. Anal. Calcd for $C_{13}H_{10}O_4$: C, 67.82; H, 4.38. Found: C, 67.54; H, 4.32.

5-(5-Formylfuran-2-yl)-2-bromobenzoic Acid Hemihydrate (**8i**). Brown microcrystals, mp 161 °C (dec); ¹H NMR (DMSO- d_6) δ 9.65 (s, 1H), 8.18 (d, J = 2.1 Hz, 1H), 7.90–7.87 (m, 2H), 7.68 (d, J = 3.6 Hz, 1H), 7.46 (d, J = 3.6 Hz, 1H); ¹³C NMR (DMSO- d_6) δ 178.4, 167.0, 156.2, 152.1, 134.9, 128.4, 128.2, 126.6, 125.4, 121.1, 110.3. Anal. Calcd for C₁₂H₇-BrO₄· 1 /₂H₂O: C, 47.40; H, 2.65. Found: C, 47.17; H, 2.52.

3-(5-Formylfuran-2-yl)-2-chlorobenzoic Acid (8j). Yellowish microcrystals, mp 192–195 °C; ¹H NMR (DMSO- d_6) δ 9.69 (s, 1H), 8.00 (dd, J = 1.5, 7.8 Hz, 1H), 7.76 (dd, J = 1.5, 7.5 Hz, 1H), 7.70 (d, J = 3.9 Hz, 1H), 7.59 (t, J = 7.8 Hz, 1H), 7.40 (d, J = 3.6 Hz, 1H); ¹³C NMR (DMSO- d_6) δ 178.6, 167.3, 154.1, 151.9, 135.3, 131.3, 130.5, 128.7, 128.1, 127.9, 124.5, 114.3. Anal. Calcd for $C_{12}H_7ClO_4$: C, 57.51; H, 2.82. Found: C, 57.51; H, 2.85.

3-(5-Formylfuran-2-yl)-2-fluorobenzoic Acid (**8k**). Orange microcrystals, mp 260–262 °C; ¹H NMR (DMSO- d_6) δ 9.69 (s, 1H), 8.15–8.10 (m, 1H), 7.96–7.91 (m, 1H), 7.72 (d, J = 3.7 Hz, 1H), 7.47 (t, J = 7.8 Hz, 1H), 7.22 (t, J = 3.8 Hz, 1H); ¹³C NMR (DMSO- d_6) δ 178.3, 164.6 (d, J = 2.9 Hz, 1C), 157.7 (d, J = 264.5 Hz, 1C), 151.6, 151.6 (d, J = 2.3 Hz, 1C), 132.7 (d, J = 1.4

Hz, 1C), 131.0 (d, J=2.9 Hz, 1C), 125.1 (d, J=4.6 Hz, 1C), 125.0 (br s, 1C), 120.8 (d, J=10.3 Hz, 1C), 118.1 (d, J=12.0 Hz, 1C), 113.3 (d, J=12.0 Hz, 1C). Anal. Calcd for C₁₂H₇FO₄: C, 61.54; H, 3.01. Found: C, 61.53; H, 2.99.

 38.43 ± 2.22

 38.37 ± 2.71

 5.53 ± 0.08

 78.92 ± 4.66

 0.442 ± 0.039

 0.598 ± 0.155

 0.353 ± 0.009

 4.043 ± 0.079

3-(5-Formylfuran-2-yl)nicotinic Acid (8l). Orange microcrystals, mp 287-289 °C; ¹H NMR (DMSO- d_6) δ 9.69 (s, 1H), 9.31 (d, J=2.1 Hz, 1H), 9.08 (d, J=1.8 Hz, 1H), 8.62 (t, J=2.1 Hz, 1H), 7.73 (d, J=3.9 Hz, 1H), 7.61 (d, J=3.9 Hz, 1H); ¹³C NMR (DMSO- d_6) δ 178.5, 165.9, 154.5, 152.6, 150.5, 149.7, 132.5, 127.1, 125.1, 125.0, 111.1. Anal. Calcd for C₁₁H₇NO₄: C, 60.83; H, 3.25; N, 6.45. Found: C, 60.47; H, 3.37; N, 6.14.

2-Chloro-5-(5-formylfuran-2-yl)nicotinic Acid (8m). Orange microcrystals, mp 215–217 °C; NMR (DMSO- d_6) δ 9.68 (s, 1H), 9.01 (d, J = 2.5 Hz, 1H), 8.56 (d, J = 2.3 Hz, 1H), 7.71 (d, J = 3.8 Hz, 1H), 7.58 (d, J = 3.7 Hz, 1H); 13 C NMR (DMSO- d_6) δ 178.6, 165.5, 153.7, 152.6, 147.9, 147.4, 135.4, 129.5, 125.0, 124.5, 111.5. Anal. Calcd for C₁₁H₆ClNO₄: C, 52.51; H, 2.40; N, 5.57. Found: C, 52.12; H, 2.58; N, 4.97.

2-Bromo-5-(5-formylfuran-2-yl)nicotinic Acid (8n). Yellow microcrystals; mp 200–210 °C (dec); ¹H NMR ((CD₃)₂CO) δ 9.81 (s, 1H), 9.07 (d, J=2.4 Hz, 1H), 8.65 (d, J=2.4 Hz, 1H), 7.67 (d, J=3.6 Hz, 1H), 7.55 (d, J=3.6 Hz, 1H); ¹³NMR ((CD₃)₂CO) δ 178.5, 165.9, 154.6, 154.2, 148.7, 140.5, 135.9, 131.6, 125.8, 124.2, 111.7. Anal. Calcd for C₁₁H₆BrNO₄: C, 44.62; H, 2.04; N, 4.73. Found: C, 45.31; H, 2.23; N, 4.24.

4-Methyl-5-(5-formylfuran-2-yl)nicotinic Acid Hemihydrate (80). Yellow microcrystals; mp 180 °C (dec); ¹H NMR ((CD₃)₂CO) δ 9.76 (s, 1H), 9.03 (s, 1H), 8.96 (s, 1H), 7.62 (d, J = 3.6 Hz, 1H), 7.15 (d, J = 3.9 Hz, 1H), 2.75 (s, 3H); ¹³NMR ((CD₃)₂CO) δ 178.5, 167.6, 156.0, 154.1, 152.4, 152.0, 147.1, 128.7, 127.6, 123.6, 114.4, 18.1. Anal. Calcd for C₁₂H₉NO₄· ¹/₂H₂O: C, 60.00; H, 4.20; N, 5.83. Found: C, 60.07; H, 3.86; N, 5.26.

General Procedure for Preparation of 2-Aryl 5-(4-Oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans (11a-o). A mixture of 3-(5-formylfuran-2-yl)benzoic acid 8 (0.75 mmol), 3-phenethyl-2-thioxothiazolidin-4-one 6 (0.18 g, 0.75 mmol), and 2 drops of 2,2,6,6-tetramethylpiperidine in 7 mL of ethanol was heated under reflux for 2.0-12.0 h (reaction time is given in Table 2). After the mixture was cooled to 40-45 °C, the solid obtained was filtered and washed with cooled ethanol to give the corresponding 2-aryl 5-(4-oxo-3-phenethyl-2-thioxothiazolidinylidenemethyl)furans (11a-o).

3-{5-[4-Oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]-furan-2-yl}benzoic Acid (11a). Yellow needles, mp 263–265 °C; 1 H NMR (DMSO- d_6) δ 8.38 (s, 1H), 8.05 (d, J = 7.6 Hz, 1H), 7.97 (d, J = 7.6 Hz, 1H), 7.68 (t, J = 7.7 Hz, 1H), 7.63 (s, 1H), 7.44 (d, J = 3.2 Hz, 1H), 7.36 (d, J = 3.3 Hz, 1H), 7.30–7.23 (m, 5H),

4.22 (t, J = 7.1 Hz, 2H), 2.95 (t, J = 7.0 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 193.7, 174.5, 166.7, 166.3, 156.9, 149.5, 137.6, 131.9, 129.8, 128.9, 128.7, 128.5, 128.4, 126.6, 124.9, 123.0, 118.9, 118.2, 111.0, 45.2, 32.1. Anal. Calcd for C₂₃H₁₇NO₄S₂: C, 63.43; H, 3.93; N, 3.22. Found: C, 63.21; H, 3.80; N, 3.20.

2-Methyl-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11b). Yellow microcrystals, mp 297–298 °C; ¹H NMR (DMSO- d_6) δ 8.27 (d, J = 1.8 Hz, 1H), 7.88 (dd, J = 7.8, 1.8 Hz, 1H), 7.63 (s, 1H), 7.49 (d, J =8.1 Hz, 1H), 7.34–7.21 (m, 7H), 4.23 (t, J=7.7 Hz, 2H), 2.96 (t, J=7.6 Hz, 2H), 2.57 (s, 3H); ¹³C NMR (DMSO- d_6) δ 193.7, 168.1, 166.3, 157.2, 149.2, 140.2, 137.6, 132.7, 131.5, 128.7, 128.5, 127.2, 126.6, 126.4, 125.9, 123.2, 118.5, 118.3, 110.4, 45.3, 32.2, 21.2. Anal. Calcd for C₂₄H₁₉NO₄S₂: C, 64.12; H, 4.26; N, 3.12. Found: C, 63.85; H, 4.21; N, 3.02.

2-Ethyl-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11c). Orange microcrystals, mp 251–253 °C; ¹H NMR (DMSO- d_6) δ 8.22 (d, J = 1.92 Hz, 1H), 7.91 (dd, J = 8.1, 1.92 Hz, 1H), 7.63 (s, 1H), 7.51 (d, J = 8.1 Hz, 1H), 7.36-7.29 (m, 4H), 7.24-7.21 (m, 3H), 4.22 (t, J=7.4 Hz, 2H), 3.00-2.93 (m, 4H), 1.20 (t, J = 7.6 Hz, 3H); 13 C NMR (DMSO- d_6) δ 193.7, 168.3, 166.3, 157.2, 149.3, 145.9, 137.6, 131.5, 131.3, 128.7, 128.5, 127.2, 126.6, 126.3, 125.9, 123.2, 118.5, 118.3, 110.4, 45.2, 32.2, 26.7, 15.8. Anal. Calcd for C₂₅H₂₁NO₄S₂: C, 64.77; H, 4.57; N, 3.02. Found: C, 64.66; H, 4.48; N, 3.04.

2-Chloro-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11d). Orange microcrystals, mp 276–278 °C; ¹H NMR (DMSO- d_6) δ 8.21 (d, J = 2.2 Hz, 1H), 7.93 (dd, J = 8.5, 2.2 Hz, 1H), 7.75 (d, J = 8.5 Hz, 1H), 7.64 (s, 1H), 7.48 (d, J = 3.7 Hz, 1H), 7.37 (d, J = 3.8 Hz, 1H), 7.33-7.29 (m, 2H), 7.25-7.21 (m, 3H), 4.23 (t, J=7.1 Hz, 2H), 2.96 (t, J = 7.1 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 193.6, 166.3, 166.3, 155.9, 149.8, 137.6, 132.6, 131.9, 131.8, 128.7, 128.6, 127.6, 127.5, 126.7, 126.3, 123.0, 119.2, 118.1, 111.6, 45.3, 32.2. Anal. Calcd for C₂₃H₁₆ClNO₄S₂: C, 58.78; H, 3.43; N, 2.98. Found: C, 59.04; H, 3.28; N, 2.95.

2-Fluoro-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11e). Yellow microcrystals, mp 286–288 °C; ¹H NMR (DMSO- d_6) δ 8.29 (dd, J = 6.7, 2.2Hz, 1H), 8.07-8.02 (m, 1H), 7.62 (s, 1H), 7.53 (dd, J = 10.4, 8.9Hz, 1H), 7.40 (d, J = 3.8 Hz, 1H), 7.35 (d, J = 3.8 Hz, 1H), 7.31-7.21 (m, 5H), 4.22 (t, J = 7.6 Hz, 2H), 2.95 (t, J = 7.6 Hz, 2H); 13 C NMR (DMSO- d_6) δ 193.6, 166.3, 164.4 (d, J = 3.4 Hz, 1C), 161.1 (d, J = 260.5 Hz, 1C), 156.1, 149.5, 137.6, 130.2 (d, J =9.2 Hz, 1C), 128.7, 128.5, 127.7, 126.6, 125.2 (d, J = 3.4 Hz, 1C), 123.1, 120.3 (d, J=11.5 Hz, 1C), 118.8, 118.4 (d, J=23.5 Hz, 1C),118.2, 110.8, 45.2, 32.1. Anal. Calcd for C₂₃H₁₆FNO₄S₂: C, 60.91; H, 3.56; N, 3.09. Found: C, 60.90; H, 3.38; N, 3.06.

2-Hydroxy-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11f). Orange microcrystals, mp 257–259 °C; ¹H NMR (DMSO- d_6) δ 8.24 (d, J = 2.2 Hz, 1H), 8.00 (dd, J = 8.7, 2.2 Hz, 1H), 7.59 (s, 1H), 7.34–7.21 (m, 7H), 7.14 (d, J = 8.7 Hz, 1H), 4.22 (t, J = 7.3 Hz, 2H), 2.95 (t, J =7.3 Hz, 2H). Anal. Calcd for C₂₃H₁₇NO₅S₂: C, 61.18; H, 3.79; N, 3.10. Found: C, 61.02; H, 3.75; N, 3.09.

2-Methoxy-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11g). Orange microcrystals, mp 221–223 °C; ¹H NMR (DMSO- d_6) δ 8.14 (d, J =2.3 Hz, 1H), 7.99 (dd, J = 8.8, 2.3 Hz, 1H), 7.64 (s, 1H), 7.40-7.31 (m, 5H), 7.28-7.24 (m, 3H), 4.26 (t, J = 7.3 Hz, 2H), 3.93 (s, 3H), 2.98 (t, J = 8.0 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 193.6, 166.8, 166.3, 158.6, 157.5, 148.8, 137.7, 128.9, 128.7, 128.5, 126.8, 126.6, 123.5, 122.3, 120.8, 118.3, 117.7, 113.5, 109.4, 56.1, 45.2, 32.2. Anal. Calcd for C₂₄H₁₉NO₅S₂: C, 61.92; H, 4.11; N, 3.01. Found: C, 61.68; H, 4.02; N, 3.02.

2-Methyl-3-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11h). Red microcrystals, mp 248-250 °C; ¹H NMR (DMSO- d_6) δ 7.85 (d, J = 7.6 Hz, 1H), 7.78 (d, J=7.7 Hz, 1H), 7.69 (s, 1H), 7.50 (t, J=7.8 Hz, 1H), 7.42 (d, J = 3.6 Hz, 1H), 7.31–7.24 (m, 5H), 7.13 (d, J = 3.6 Hz, 1H), 4.24 (t, J=7.8 Hz, 2H), 2.96 (t, J=8.2 Hz, 2H), 2.61 (s, 3H); ¹³C NMR (DMSO- d_6) δ 193.7, 166.3, 157.3, 149.2, 137.6, 135.0, 130.4, 130.0, 128.7, 128.5, 126.6, 126.3, 122.7, 118.8, 118.4, 114.5, 45.3, 32.1, 18.5. Anal. Calcd for C₂₄H₁₉NO₄S₂: C, 64.12; H, 4.26; N, 3.12. Found: C, 63.75; H, 4.42; N, 3.14.

nemethyl]furan-2-yl}benzoic Acid (11i). Deep-red microcrystals; mp 241 °C (dec); ¹H NMR (DMSO- d_6) δ 8.20 (d, J=2.1 Hz, 1H), 7. 94–7.98 (m, 1H), 7.86–7.89 (m, 1H), 7.69 (s, 1H), 7.54 (d, J=3.6 Hz, 1H), 7.41 (d, J = 3.6 Hz, 1H), 7.24–7.38 (m, 6H), 4.27 (t, $J=8.4 \text{ Hz}, 2\text{H}), 3.00 \text{ (t, } J=7.5 \text{ Hz}, 2\text{H}); {}^{13}\text{C NMR (DMSO-}d_6) \delta$ 193.7, 167.1, 166.4, 156.0, 149.8, 137.7, 135.1, 135.0, 128.8, 128.6, 128.0, 127.5, 126.7, 126.0, 123.1, 120.4, 119.3, 118.2, 111.7, 45.4, 32.2. HRMS (DIP-CI-MS) calcd for C₂₃H₁₆-BrNO₄S₂: [M + H] 513.9779. Found: 513.9782. Anal. HPLC: 98.8%, $t_R = 3.38$ (tracing, 0.7%, $t_R = 4.19$).

2-Chloro-3-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11j). Orange microcrystals, mp 258-260 °C; ¹H NMR (DMSO- d_6) δ 8.0 (dd, J=7.8, 1.5 Hz, 1H), 7.75-7.64 (m, 3H), 7.47 (d, J = 3.6 Hz, 1H), 7.40 (d, J =3.6 Hz, 1H), 7.33-7.29 (m, 2H), 7.24-7.21 (m, 3H), 4.23 (t, J=7.5 Hz, 2H), 2.96 (t, J = 7.5 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 193.9, 167.4, 166.5, 154.0, 149.7, 137.8, 135.7, 130.3, 130.0, 128.9, 128.7, 128.5, 128.1, 127.4, 126.8, 122.5, 120.0, 118.3, 115.9, 45.5, 32.3. Anal. Calcd for C₂₃H₁₆ClNO₄S₂: C, 58.78; H, 3.43; N, 2.98. Found: C, 58.40; H, 3.39; N, 2.88.

2-Fluoro-3-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}benzoic Acid (11k). Orange microcrystals, mp 262-264 °C; ¹H NMR (DMSO- d_6) δ 8.00-7.95 (m, 1H), 7.87 - 7.81 (m, 1H), 7.65 (s, 1H), 7.50 (t, J = 7.7 Hz, 1H), 7.39 (d, J = 3.7 Hz, 1H, 7.33 - 7.29 (m, 2H), 7.25 - 7.18 (m, 4H), 4.22 (t, 2H)J = 8.0 Hz, 2H), 2.95 (t, J = 8.0 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 193.7, 166.3, 164.9, 157.2 (d, J = 263.4 Hz, 1C), 151.6, 149.3, 137.6, 132.0, 129.4 (br s, 1C), 128.7, 128.6, 126.7, 125.1 (d, *J* = 4.6 Hz, 1C), 122.9, 122.2 (br d, J = 10.9 Hz, 1H), 119.6, 118.0, 117.9 (d, J = 14.9 Hz, 1C), 114.6 (d, J = 13.2 Hz, 1C), 45.3, 32.1. Anal. Calcd for C₂₃H₁₆FNO₄S₂: C, 60.91; H, 3.56; N, 3.09. Found: C, 61.13; H, 3.52; N, 3.11.

5-{5-[4-Oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}nicotinic Acid (111). Orange microcrystals, mp 259–261 °C; ¹H NMR (DMSO- d_6) δ 9.14 (d, J = 2.2 Hz, 1H), 9.03 (d, J = 1.9 Hz, 1H), 8.58 (t, J = 2.1 Hz, 1H), 7.67 (s, 1H), 7.56 (d, J = 3.7 Hz, 1H), 7.40 (d, J = 3.7 Hz, 1H), 7.34–7.22 (m, 5H), 4.24 (t, J = 7.7 Hz, 2H), 2.97 (t, J = 7.7 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 193.6, 166.3, 166.2, 154.9, 150.2, 150.0, 147.3, 137.6, 131.6, 130.9, 128.7, 128.5, 126.6, 124.4, 122.8, 119.4, 118.1, 111.8, 45.3, 32.1. Anal. Calcd for $C_{22}H_{16}N_2O_4S_2$: C, 60.54; H, 3.69; N, 6.42. Found: C, 60.44; H, 3.65; N, 6.28.

2-Chloro-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}nicotinic Acid (11m). Orange microcrystals, mp 246–248 °C; ¹H NMR (DMSO- d_6) δ 8.71 (d, J = 2.3 Hz, 1H), 8.12 (d, J = 2.5 Hz, 1H), 7.66 (s, 1H), 7.51 (d, J = 3.9 Hz, 1H), 7.38 (d, J = 3.7 Hz, 1H), 7.33–7.29 (m, 2H), 7.24–7.22 (m, 3H), 4.23 (t, J = 7.3 Hz, 2H), 2.96 (t, J = 7.3 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 193.6, 166.6, 166.3, 154.5, 149.9, 146.4, 143.1, 138.3, 137.6, 132.1, 128.7, 128.6, 126.7, 123.8, 122.8, 119.4, 118.1, 11.9, 45.3, 32.2. Anal. Calcd for C₂₂H₁₅ClN₂O₄S₂: C, 56.11; H, 3.21; N, 5.95. Found: C, 56.01; H, 3.01; N, 5.60.

2-Bromo-5-{5-[4-oxo-3-phenethyl-2-thioxothiazolidin-5-ylidenemethyl]furan-2-yl}nicotinic Acid Hemi 2,2,6,6-Tetramethylpiperidine Dihydrate (11n). Orange microcrystals; mp 200-205 °C; ¹H NMR (DMSO- d_6) δ 8.79 (d, J = 2.4 Hz, 1H), 8.41 (br s, 1H), 8.25 (d, J = 2.4 Hz, 1H), 7.64 (s, 1H), 7.55 (d, J = 3.9Hz, 1H), 7.37 (d, J = 3.9 Hz, 1H), 7.33–7.21 (m, 5H), 4.22 (t, $J=7.2 \text{ Hz}, 2\text{H}), 2.95 \text{ (t, } J=7.8 \text{ Hz}, 2\text{H}); ^{13}\text{C NMR (DMSO-}d_6) \delta$ 193.6, 166.7, 166.3, 153.8, 150.2, 145.3, 138.1, 137.6, 132.6, 128.7, 128.6, 126.7, 124.2, 122.7, 119.8, 118.0, 112.6, 55.6, 45.3. Anal. Calcd for $C_{22}H_{15}S_2BrN_2O_4 \cdot \frac{1}{2}C_9H_{19}N \cdot 2H_2O$: C, 52.69; H, 4.42; N, 5.80. Found: C, 52.84; H, 4.02; N, 5.32.

Molecular Modeling and Docking of 11d onto the gp41 Hydrophobic Cavity. We used the automated docking software Glide (Schrodinger, Portland, OR). This applies a two-stage scoring process to sort out the best conformations and orientations of the ligand (defined as pose) based on its interaction pattern with the receptor. The starting point of the docking simulation was the X-ray structure of the gp41 core (1AIK) available from the Protein Data Bank (PDB) at the "Research Collaboratory for Structural Bioinformatics (RCSB)"; this has been extensively used in our research program. Three-dimensional coordinates of the ligands and their isomeric, ionization, and tautomeric states were generated using the LigPrep (including Ionizer) module from Schrodinger. The protein was prepared using the protein preparation tool available in the software. A grid file encompassing the area in the cavity that contains information on the properties of the associated receptor was created. Conformational flexibility of the ligands was handled via an exhaustive conformational search. We used Schrodinger's proprietary GlideScore scoring function in standard precision (SP) and extra precision (XP) mode to score the optimized poses. The poses were selected on the basis of the salt bridge and other hydrophobic and hydrogen bond interactions.

Determination of the Inhibitory Activity of the Compounds on HIV-1 Replication. The inhibitory activity of compounds on HIV-1_{IIIB} replication in MT-2 cells was determined as previously described. 18 In brief, 1×10^4 MT-2 cells were infected with an HIV-1_{IIIB} strain (100 TCID₅₀) in 200 μ L of RPMI 1640 medium containing 10% FBS in the presence or absence of a test compound at graded concentrations overnight. Then the culture supernatants were removed and fresh media containing no test compounds were added. On the fourth day postinfection, an amount of 100 µL of culture supernatants was collected from each well, mixed with equal volumes of 5% Triton X-100, and assayed for p24 antigen, which was quantitated by ELISA. Briefly, wells of polystyrene plates (Immulon 1B, Dynex Technology, Chantilly, VA) were coated with HIV immunoglobulin (HIVIG), which was prepared from plasma of HIV-seropositive donors with high neutralizing titers against $HIV-1_{IIIB}$, in 0.085 M carbonate-bicarbonate buffer (pH 9.6) at 4 °C overnight, followed by washing with buffer (0.01 M PBS containing 0.05% Tween-20) and blocking with PBS containing 1% dry fatfree milk (Bio-Rad, Inc., Hercules, CA). Virus lysates were added to the wells and incubated at 37 °C for 1 h. After extensive washes, anti-p24 mAb (183-12H-5C), biotin-labeled antimouse IgG1 (Santa Cruz Biotechnolgy, Santa Cruz, CA), streptavidinlabeled horseradish peroxidase (Zymed, S. San Francisco, CA), and the substrate 3,3',5,5'-tetramethylbenzidine (Sigma Chemical Co., St. Louis, MO) were added sequentially. Reactions were terminated by addition of 1 N H₂SO₄. Absorbance at 450 nm was recorded in an ELISA reader (Ultra 386, TECAN, Research Triangle Park, NC). Recombinant protein p24 purchased from United States Biological (Swampscott, MA) was included for establishing standard dose-response curves. Each sample was tested in triplicate. The percentage of inhibition of p24 production was calculated as previously described. 31 EC $_{50}$ values were calculated using the computer program CalcuSyn,³² kindly provided by Dr. T. C. Chou (Sloan-Kettering Cancer Center, New York).

To determine the potential effect of light on the bioactivity of the compounds, we also tested these compounds under exclusion of light. Briefly, the compound solutions were prepared and added to the virus (HIV- $l_{\rm IIIB}$) and MT-2 cells in the biosafety hood with the light turned off. Then the plates were wrapped with foil before they were put in the CO_2 incubator. The other procedures are the same as described above.

Inhibitory activity of compounds on infection by a primary HIV-1 isolate, 94UG103 (X4R5, clade A), was determined as previously described. 18 Peripheral blood mononuclear cells (PBMCs) were isolated from the blood of healthy donors at the New York Blood Center by standard density gradient centrifugation using Histopaque-1077 (Sigma). The cells were plated in 75 cm² plastic flasks and incubated at 37 °C for 2 h. The nonadherent cells were collected and resuspended at 5×10^6 in 10 mL of RPMI-1640 medium containing 10% FBS, 5 μg/mL PHA, and 100 U/mL IL-2 (Sigma), followed by incubation at 37 °C for 3 days. The PHA-stimulated cells were infected with 94UG103 at 0.01 multiplicity of infection (MOI) in the absence or presence of a compound at graded concentrations. Culture media were changed every 3 days. The supernatants were collected 7 days postinfection and tested for p24 antigen by ELISA as described above. The percent inhibition of p24 production and EC_{50} values were calculated as described above.

Assessment of in Vitro Cytotoxicity. The in vitro cytotoxicity of compounds on MT-2 cells was measured by XTT assay. Briefly, an amount of $100~\mu\text{L}$ of the test compound at graded concentrations was added to equal volumes of cells ($5 \times 10^5/\text{mL}$) in wells of 96-well plates. After incubation at 37 °C for 4 days, $50~\mu\text{L}$ of XTT solution (1 mg/mL) containing $0.02~\mu\text{M}$ phenazine methosulfate (PMS) was added. After 4 h, the absorbance at 450 nm was measured with an ELISA reader. The CC₅₀ (concentration for 50% cytotoxicity) values were calculated using the CalcuSyn program. 32

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Supporting Information Available: Experimental details, spectral data for aryl(heteryl) bromides/iodides **2c**, **2j**, **2n**, **2o**, intermediates **3n**, **3o**, **4o**, 3-phenethyl-2-thioxothiazolidin-4-one **6**, and the inhibition of compounds **11a**, **11b**, and **11d** on infection by HIV-1_{IIIB} and 94UG103. This material is available free of charge via the Internet at http://pubs.acs.org.

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